A NEW DISACCHARIDE, GLAUCOBIOSE FROM CHINESE DRUG "PAI-CH'IEN": A COMPARISON OF

13C NMR WITH ITS DIASTEREOMERIC ISOMER, STROPHANTHOBIOSE

Takashi Nakagawa, Koji Hayashi, Keiji Wada, and Hiroshi Mitsuhashi*

Faculty of Pharmaceutical Sciences, Hokkaido University,

Sapporo 060, Japan

Abstract: The structure of a new disaccharide named glaucobiose (4) has been established, and the 13 C NMR of its methyl β - (5) and α -glycoside (6), and that of methyl β -strophanthobioside (2) were studied.

We have already reported the structures of aglycones and glycosides obtained from "Pai-ch'ien" ^{2a)} ("Bai-qian" ^{2b)}), dried root of <u>Cynanchum glaucescens Hand-Mazz</u> (Asclepiadaceae). We wish to report now the structure of a new disaccharide named glaucobiose (4) and the ¹³C NMR study on glycosidation shifts for its methyl β - (5) and α -glycoside (6), and methyl β -strophanthobioside (2).

The rather polar fractions of crude glycosides of this drug were hydrolyzed with 75% MeOH-0.05 N H₂SO₄ at 50°, 30 min, and the hydrolysates subjected to repeated silica gel column chromatography to yield $\underline{5}$ and $\underline{6}$ in the ratio 5:2 as colorless fine needles. Compound $\underline{5}$, mp 95-98°, [α]_D-32.1° (\underline{c} =1.00, MeOH), and $\underline{6}$, mp 143-145°, [α]_D -156° (\underline{c} =1.00, MeOH) possessed the same molecular formulae of $C_{14}H_{26}O_9$ on the bases of their elemental analyses and FD-MS ($\underline{m}/\underline{z}$: 339(M+H) †). Each of them was further hydrolyzed with 0.05 NH₂SO₄ at 50°, 30 min to give glaucobiose (4) as an amorphous hygroscopic white powder, [α]_D -72.6° (\underline{c} =0.81, H₂O), $C_{13}H_{24}O_9$ ($\underline{m}/\underline{z}$: 325(M+H) †), which showed positive Keller-Kiliani reaction suggesting the presence of 2-deoxysugar. On hydrolysis using snail enzyme (β -glucosidase) $\underline{4}$ afforded cymarose and glucose. The 1 H NMR (Table I) of $\underline{5}$ and $\underline{6}$ showed that they are the anomeric isomers of methyl glycosides. This was further established by the J-values of $^{13}C_-^{-1}H$ coupling on the anomeric centers of them 3), 155.2 Hz for the glucose and 158.1 Hz for the cymarose in $\underline{5}$, and 158.8 Hz for the glucose and 166.2 Hz for the cymarose in 6.

On acetylation $\underline{4}$ gave penta-O-acetyl glaucobiose (7), $163-165.5^{\circ}$, [α]_D -33.2° (\underline{c} =1.71, CHCl₃), whose $^{1}\text{H NMR}$ (400 MHz) in CDCl $_{3}$ revealed all the proton signals without overlapping. assignments 4) are listed in Table II. This was supported by the prominent fragment peaks at m/z: 145 (base peak) and 163 in the FD-MS of 4 (Fig. 1). On the other hand, a well known disaccharide strophanthobiose (1) $(4-O-\beta-\underline{D}-glucopyranosyl-\underline{D}-cymaropyranose)^5$ was identified from C. caudatum Max (Asclepiadaceae).

Table I. 1 H NMR data for 5 and 6 (δ in pyridine-d₅) (200 MHz)

	<u>5</u>	6_
6-CH3	1.49 (3H, d, J=6.3 Hz)	1.41 (3H, d, J=6.6 Hz)
	1.68 (1H, $dd\overline{d}$, J=13.5, 9, 2 Hz)	1.69 (1H,ddd, J=14,4, 3.5 H
2-CHe	2.30 (1H, ddd, \overline{J} =13.5, 4, 2 Hz)	2.30 (1H, ddd, \overline{J} =14,4,2 Hz)
	3.46, 3.56 (each 3H, s)	3.30, 3.47 (each 3H, s)
	4.90 (1H, dd, J=9, 2 Hz)	4.67 (1H, dd, J=4, 2 Hz)
	5.00 (1H, d, J=7.6 Hz)	4.99 (1H, d, J=7.6 Hz)

a=axial, e=equatorial

Chart 1

Table II. 1 H NMR data for $\frac{7}{2}$ (δ in CDC1₃)(400 MHz)

```
m/z: 163
                                                 1-CHa
                                                            6.00(1H,dd, J=9, 2.4 Hz)
                                                                                                                         1'-CHa
                                                                                                                                    4.67(1H,d,J=7.8 Hz)
 ҪН₂ОН
                                                            1.66 (lH,ddd, J=13.7, 9,3 Hz)
2.24 (lH,ddd, J=13.7, 5.4, 2.4 Hz)
3.75 (lH,ddd, J=5.4, 3,3 Hz)
                                                                                                                        2'-CHa 4.98 (1H, dd, J=9.8, 7.8 Hz) 3'-CHa 5.08 (1H, dd, J=9.8, 9.3 Hz) 4'-CHa 5.22 (1H, dd, J=9.8, 9.3 Hz)
                                                 2-CHa
                                                 2-CHe
                                                 3-CHe
                                                                                                                                    3.69 (1H, ddd, J=9.8, 5.4, 2.4 Hz)
                                                 3-00H<sub>3</sub>
                                                            3.39(3H,s)
                                                                                                                         5'-CHa
                                                                                                                                   4.14 (1H,dd, J=12, 2.4 Hz)
4.21 (1H,dd, J=12, 5.4 Hz)
                                                            3.51(1H,dd, J=8.3, 3 Hz)
                   145 (base péak)
                                                5-CHa 4.08(1H,dq, \overline{J}=8.3, 6.4 Hz)
5-CH<sub>3</sub> 1.27(3H,d, \overline{J}=6.4 Hz)
         Fig. 1
                                                                                                                                    2.01, 2.03 \times 2, 2.08 \times 2 (each 3H,s)
```

a=axial, e=equatorial

Tanaka et al.⁶⁾ and Tori et al.⁷⁾ reported the β -D-glucosidation effects on carbon chemical shifts (glucosidation shifts) of both glucose and aglycone moieties. According to their studies, when one of the β-carbons of aglycone is substituted, the magnitude of glucosidation shifts depends significantly on the position, syn or anti to the pyranose-ring oxygen atom in the most stable conformation averaged around the glycosidic linkage.

Table III.	¹³ c	NMR	Chemical	Shifts	for	2, 5.	6.	8.	9.	and	10
			(ppm in	pyridine	e-d _E)	(25)	ηĤz՝	, ·	_,		

	<u>8</u>	10	2	5	9	<u>6</u>
C-1	99.4		99.3	99.2	97.6	97.5
-2	35.1		36.1	35.2	31.9	31.9
-2 -3	78.5		77.8(-0.7) ^b	78.2 ^a)	76.5	73.2(-3.3)
-4 -5	74.0		83.2(+9.2)	79.0(+5.0)	73.2	78.2(+5.0)
- 5	71.0		69.4(-1.6)	69.3(-1.7)	65.2	64.2(-1.0)
-6	18.9		18.6	18.9	18.5	18.3
-OMe	56.0		56.0	55.8	56.7	56.7
	57.8		58.4	57.9	54.7	54.8
C-1' -2' -3' -4' -5' -6'		105.4	106.5(+1.1)	101.8(-3.6)		101.7(-3.7)
-2'		74.8	75.4	74.9 78.4 ^a)		75.1
-3′		78.1	78.4	78.4 ^{d)}		78.4
-4		71.4	71.9	71.6 78.2 ^a)		71.9
-5′		78.1	78.4	78.2 ^a)		78.4
-6′		62.5	63.1	62.7		62.8
-0Me		56.7				

- a) Assignments may be interchanged. b) Values (in parentheses) are $\Delta \delta = \delta_2 \delta_{g(0)}$.

Namely, alkyl-substitution on the $syn-\beta$ -carbon in a secondary alcohol, the anomeric carbon chemical shift of glucose (C-1') changes ca. -4 ppm from that of methyl g-D-glucopyranoside (10)8) and the α -carbon of aglycone (C- α) +5.5 (±1.5 ppm), while on the anti- β -carbon, the C-1 changes 0 (± 1.5 ppm) from that of 10 and the C- α +10.4 (± 1.5 ppm). The 13 C NMR for 2, 5, and 6 were measured and their carbon chemical shifts assigned (Table III) on the bases of methyl β -D-cymaropyranoside (8) 4a , $^{9)}$, methyl α - $\underline{\underline{\underline{}}}$ -cymaropyranoside (9) 1b , 4a , and $\underline{\underline{10}}^{8)}$.

HOOH

$$CH_3$$
 OH
 OH

In the 13 C NMR of 13 C (Table III), its glucosidation shifts pattern is evidently corresponding to the case of anti- β -substitution while of 13 5 and 13 6 to that of syn- β -substitution, which is illustrated in Fig. 2. As previously reported 16 1, the cymarose which constitutes the glycosides of this drug belongs to 13 5-series. Combination of these knowledge led to the structure of 13 6 as 13 6- 13 7- 13 9 glucopyranosyl- 13 9-cymaropyranose. Regarding to these three compounds, the axial-substituted C-3 methoxyl groups constributed little to the magnitude of glucosidation shifts.

From the plant taxonomical point of view, it is noteworthy that these two diastereomeric disaccharides were obtained from the same genus. To the best of our knowledge this is the first example that the glucosidation shifts could be compared between these two diastereomeric disaccarides

Acknowledgements: This work was supported in part by grants from the Ministry of Education, Japan (Grant-in-Aid, No. 447108). We are grateful to Dr. Hon-Yen Hsu for his help to obtain the drug and Mr. K. Watanabe of this University for field desorption mass spectral measurements.

References and Notes

- 1) a) T. Nakagawa, K. Hayashi, and H. Mitsuhashi, Tetrahedron Lett., 1982, 757;
 - b) T. Nakagawa, K. Hayashi, K. Wada, and H. Mitsuhashi, Tetrahedron, in press.
- 2) a) Shie Tsung-wan, Liu Mei-law, and Luo Tzu-ching, Acta Pharm. Scinica, 7, 175 (1959);
 - b) "Zhòng Yào Zhì", ed. by Chinese Medical Academy, China, 1979, p. 165.
- 3) K. Bock, I. Lundt, and C. Pedersen, Tetrahedron Lett., 1973, 1037.
- 4) a) Assignments of the ¹H NMR of <u>7</u> was based on that of reported for <u>8</u>, C. Monneret,
 C. Conreur, and Qui Khuong-Huu, <u>Carbohydr. Res.</u>, <u>65</u>, 35 (1978); and based on that for methyl β-<u>p</u>-glucopyranoside tetraacetate, K. Izumi, <u>J. Biochem.</u>, <u>7</u>, 535 (1974);
 - b) The H NMR data for 7 is nealy identical with those for penta-O-acetyl strophanthobioside (3).
- 5) a) A. Stoll, J. Renz, and W. Kreis, Helv. Chim. Acta, 20, 1484 (1937);
 - b) A. Stoll and J. Renz, Ibid., 22, 1193 (1939); c) M. Barbier and O. Schindler, Ibid., 42, 1065 (1959).
- 6) a) R. Kasai, O. Suzuo, J. Asakawa, and O. Tanaka, Tetrahedron Lett., 1977, 175;
 - b) K. Yamasaki, R. Kasai, Y. Masaki, M. Okihara, O. Tanaka, H. Oshio, S. Takagi, M.Yamaki, K. Masuda, G. Nonaka, M. Tsuboi, and I. Nishioka, Ibid., 1977, 1231;
 - c) K. Mizutani, R. Kasai, and O. Tanaka, Carbohydr. Res., 87, 19 (1980);
 - d) K. Itano, K. Yamasaki, C. Kihara, and O. Tanaka, Ibid., 87, 27 (1980).
- 7) a) K. Tori, S. Seo, Y. Yoshimura, H. Arita, and Y. Tomita, Tetrahedron Lett., 1977, 179;
 - b) S. Seo, Y. Tomita, K. Tori, and Y. Yoshimura, J. Am. Chem. Soc., 100, 3331 (1978).
- 8)a) K. Tori, S. Seo, Y. Yoshimura, M. Nakamura, Y. Tomita, and H. Ishii, <u>Tetrahedron Lett.</u>, <u>1976</u>, 4167; b) S. Yahara, K. Kasai, and O. Tanaka, <u>Chem. Pharm. Bull.</u>, <u>25</u>, 2041 (1977).
- 9) F. Abe and T. Yamauchi, Chem. Pharm. Bull., 26, 3023 (1978).

(Received in Japan 10 September 1982)